

SPECTROPHOTOMETRIC DETERMINATION OF TERBINAFINE HCl, TELMISARTAN AND RAMIPRIL THROUGH REDOX REACTIONS USING CERIC SULPHATE AND CERIC SULPHATE-CHROMATROPE 2R

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ABSTRACT

Direct (A) and indirect (B) spectrophotometric methods are proposed for the assay of Terbinafine HCl, Telmisartan and Ramipril in bulk drugs and in their dosage forms using ceric (IV) sulphate and ceric (IV) sulphate-chromatropene 2R as reagents. Method (A) involves the addition of certain amount of ceric (IV) sulphate to drugs in acid medium, followed by determination of the decrease in absorbance measured at 319 nm using the experiment as a blank.

Method (B) involves the addition of a known excess of ceric (IV) sulphate to drugs in acid medium followed by the determination of unreacted ceric by reacting with a fixed amount of chromatropene 2R then the decrease in color intensity of C2R was measured spectrophotometrically at λ_{max} 507 nm against the reagent blank.

The Beer's law is obeyed in the concentration range of (1-9) $\mu\text{g/ml}$ (A), (1-7) $\mu\text{g/ml}$ (B) of Terbinafine hydrochloride, (1-5) $\mu\text{g/ml}$ (A), (B) of Telmisartan and (10-80) $\mu\text{g/ml}$ (A), (10-60) $\mu\text{g/ml}$ (B) of Ramipril. The validity of these methods have been successfully applied for the determination of the drugs in their pharmaceutical preparations. Good recoveries were obtained and the results were comparable with those obtained by reference method.

Keywords: Terbinafine hydrochloride, Telmisartan, Ramipril, Ceric sulphate and Ceric sulphate.

INTRODUCTION

In the present work, spectrophotometric determination of Terbinafine HCl, Telmisartan and Ramipril using cerium (IV) and cerium (IV)-chromatropene 2R were described. The proposed methods are simple, accurate and in good agreement with results obtained by reference methods.

Terbinafine Hydrochloride, (TH) chemically is 1-naphthalenemethanamine, n-(6, 6-dimethyl-2-hepten-4-ynyl)-n methyl-, (E)-, hydrochloride (Figure 1). TH is a new potent antifungal agent. It belongs to an allyl amine class and has broad-spectrum activity against yeasts, dimorphic fungi, molds and dermatophytes¹⁻³. Literature survey shows several HPTLC⁴⁻⁶, non-aqueous voltametric⁷, spectrometric methods⁸⁻¹² and ion-

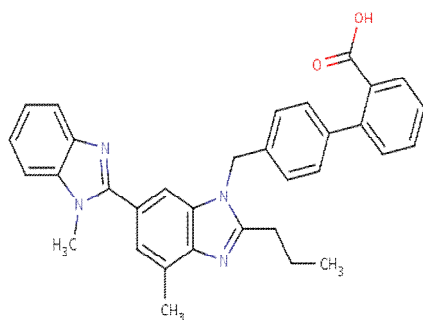
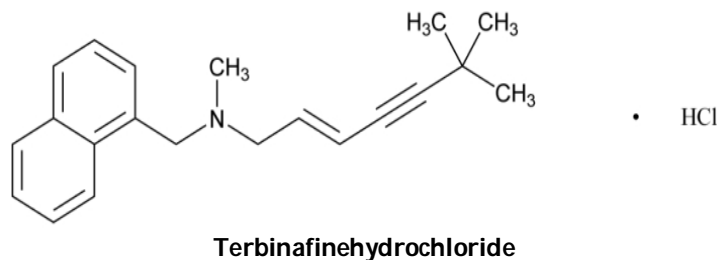
pair RP chromatography¹³ have been used for assay of TH in raw material and dosage forms. Only stability-indicating HPTLC^{12,13} method is reported for determination of the drug. Reported spectrophotometric⁹ and chromatographic^{14,15} methods estimates TH in presence of its degradant or metabolites. Also TH has been determined in biological fluids (plasma, urine) tissues, nails and cat hair by HPLC¹⁶⁻¹⁸ and in tablets and creams by HPLC^{19,20}.

Telmisartan, (TEL) chemically is 4'-[(1, 4'-dimethyl-2'-propyl [2, 6'-bi-1H-benzimidazol]-1'-yl) methyl]-[1, 1'-biphenyl]-2-carboxylic acid (Figure 1). It is indicated in the treatment of essential hypertension. The usually effective dose of Telmisartan is 20, 40 and 80 mg once

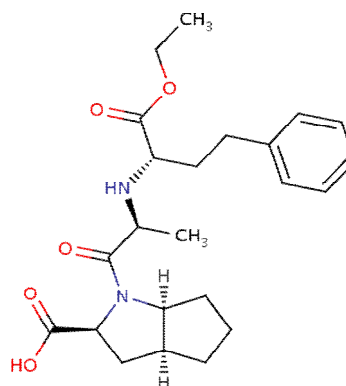
daily. Literature survey revealed that there were many methods like Spectrophotometry²¹⁻²⁶ using first order derivative²⁵. Simultaneous equation, RP-HPLC²⁷⁻³⁰ and LC-MS/MS^{31,32} and HPTLC^{33,34} for determination of Telmisartan alone and with other drugs in combination have been reported.

Ramipril, (RP) chemically is [(2S, 3aS, 6aS)-1-[(S)-2-[[[(S)-1-(ethoxycarbonyl)-3-phenylpropyl] amino] propanoyl] octahydrocyclopenta [b] pyrrole-2-carboxylic acid

(Figure 1).RP is an angiotensin-converting enzyme inhibitor (ACEI), which is widely used in the treatment of hypertension and congestive heart failure. RP plays an important role in inhibiting the conversion of the inactive angiotensin I to the active angiotensin II³⁵⁻³⁷. There are many reported methods to determine RP alone or in combination with other in dosage forms and the simultaneous determination of RP in the presence of the degradants³⁸⁻⁴⁰.



Telmisartan, TEL



Ramipril, RP

Fig. 1: Chemical structure of TH, TEL and RP

Experimental

I. Apparatus

Spectrophotometer: SHIMADZU UV-1800 PC, dual beam UV-visible spectrophotometer with two matched 1 cm quartz cells, connected to an IBM compatible personal computer (PC) and an HP-600 inkjet printer. Bundled UV-PC personal spectroscopy software version (3.7) was used to process the absorption and the derivative spectra. The spectral band width was 0.2 nm with wavelength scanning speed of 2800 nm min⁻¹.

II. MATERIALS AND METHODS

All reagents were of analytical grade and distilled water was used.

1. Terbinafine HCl (Novartis, Egypt).
2. Telmisartan (Boehringer, Egypt).
3. Ramipril (Aventis, Egypt).
4. Cerium(IV) sulphate (Merck, England), 0.15% (TEL, method A), 0.1% (TH, RAM, method A)

, 0.2 % (TH, method B), 0.3 % (TEL, method B), 0.4% (RAM, method B) w/v M solutions by dissolving 0.15, 0.1, 0.2, 0.3 0.4gm cerium(IV) sulphate in 0.5, 5, 1 M sulphuric acid.

5. Chromotrope 2R

2-(Phenylazo) chromotropic acid disodium salt, (Aldrich, Germany), 0.15 % w/v solution by dissolving 0.15 gm chromotrope 2R in distilled water.

III. Standard drug solutions

1. 0.025gm of Terbinafine HCl and Ramipril dissolved in 25ml distilled water, then further dilution with distilled water to obtain working standard solution of concentration $50\mu\text{g ml}^{-1}$.
2. 0.025 gm of Telmisartan dissolved in 25 ml 5M sulphuric acid, then further dilution with 5M sulphuric acid to obtain working standard solution of concentration $100\mu\text{g ml}^{-1}$.

IV. Pharmaceutical preparations

1. Lamisil tablets (Novartis, Egypt), labelled to contain 25mg Terbinafine hydrochloride per tablet.
2. Micardis tablets (Boehringer, Egypt), labelled to contain 8 mg Telmisartan per tablet.
3. Tritace protect tablets (Sanofiaventis, Egypt), labeled to contain 10 mg Ramipril per tablet.

V. General procedures

1- Construction of calibration curves

a. Spectrophotometric procedure using ceric(IV) sulphate

To different aliquots of standard solutions containing (1-7), (1-5) and (10-70) $\mu\text{g/ml}$ of Terbinafine HCl, Telmisartan and Ramipril, respectively. For Terbinafine HCl, 0.8 ml of 0.1 % w/v cerium(IV) sulphate in 0.5M sulphuric acid solution was added. For Telmisartan, 0.8 ml of 0.15 % w/v cerium(IV) sulphate in 5M sulphuric acid solution was added. For Ramipril, 0.6 ml of 0.1 % w/v cerium(IV) sulphate in 1M sulphuric acid solution was added. The mixtures were heated in a boiling water bath for 36, 30 and 45 min for Terbinafine HCl, Telmisartan and

Ramipril, respectively. The test tubes were then cooled and the reaction mixtures were transferred into a series of 10 ml volumetric flasks, the volume was made up to the mark with distilled water. The decrease in absorbance was measured at 319 nm using the experiment as a blank.

b. Spectrophotometric procedure using ceric (IV) sulphate and chromotrope 2R (C2R)

To different aliquots of standard solutions containing (1-7), (1-5) and (10-50) $\mu\text{g/ml}$ of Terbinafine HCl, Telmisartan and Ramipril, respectively. 1ml of 0.2% w/v cerium(IV) sulphate were added for Terbinafine HCl, 1ml of 0.3 % w/v cerium(IV) sulphate were added for Telmisartan and 1.2ml of 0.4% w/v cerium(IV) sulphate were added for Ramipril. Mixtures were heated in a boiling water bath for 20, 35 and 35 min for Terbinafine HCl, Telmisartan and Ramipril, respectively, the test tubes were then cooled and the reaction mixtures were then transferred into a series of 10 ml volumetric flasks and 1.4, 1 and 1.6 ml of C2R was added for Terbinafine HCl, Telmisartan and Ramipril, respectively, then the volume was made up to the mark with distilled water. The decrease in color intensity of C2R was measured spectrophotometrically at λ_{max} 507 nm.

2- For pharmaceutical preparations

a. Lamisil tablets

Ten tablets were powdered and an accurately weighed amount equivalent to 10 mg Terbinafine HCl was shaken with distilled water then filtered and diluted to 100 ml with distilled water to obtain working solution of concentration $100\mu\text{g ml}^{-1}$.

b. Micardis tablets

Ten tablets were powdered and a weight equivalent to 25 mg of Telmisartan was shaken with cold water for 2 min, to dissolve sorbitol, filtered, washed with 20 ml distilled water and the precipitate was transferred from the filter paper into 25 ml volumetric flask with 5M sulphuric acid then filtered and completed to the mark with 5M sulphuric acid. Further dilution was made to obtain working solution of the concentration $100\mu\text{g ml}^{-1}$ using 5M sulphuric acid.

c. Tritace Protect tablets

Ten tablets were powdered and an accurately weighed amount equivalent to 25mg Ramipril was shaken with distilled water then filtered and diluted to 25ml with distilled water, then

further dilution to obtain working solution of concentration $50\mu\text{g ml}^{-1}$.

Standard addition technique was used for analysis of the selected drugs in their commercial tablets.

VI. RESULTS AND DISCUSSION

I. Spectrophotometric procedure using ceric sulphate

Cerium (IV) sulphate being a strong oxidizing agent was used for determination of pharmaceutical compounds (41-48). The proposed method based on oxidation of the investigated drugs with excess cerium (IV) sulphate in acidic medium and subsequent measurement of the decrease in reagent absorbance at 319 nm. Figure 2, 3 and 4 show absorption spectra of reaction and blank.

Investigation of assay parameters

1. Effect of sulphuric acid concentration

Acid medium is needed to prevent precipitation of the hydrated cerium (IV) oxide ($\text{CeO}_2 \cdot x\text{H}_2\text{O}$). Using different sulphuric acid concentrations ranging from 0.5-5M solutions, the optimum concentration that gave maximum decrease in absorbance were 0.5, 5, 1M solutions for (TH, TEL and RP) respectively.

2. Effect of ceric (IV) sulphate concentration

Maximum decrease in absorbance was obtained using 0.8 ml 0.1% w/v ceric (IV) sulphate, for Terbinafine HCl, 0.8 ml 0.15% w/v ceric (IV) sulphate, for Telmisartan and 0.6 ml 0.1% w/v ceric (IV) sulphate, for Ramipril.

3. Effect of temperature and heating time

Heating for 25 min in a boiling water bath was optimum for Terbinafine HCl, Telmisartan and 45 min in a boiling water bath was optimum for Ramipril.

II. Spectrophotometric procedure using ceric sulphate and chromotrope 2R

The proposed method involves two stages: the first one is the oxidation of the selected drugs with known excess Ce^{+4} in acidic medium under the effect of heating as explained in the previous paragraph and the second one involves the determination of the unreacted oxidant by measuring the decrease in absorbance of C2R at the suitable λ_{max} 507 nm. Figure 5, 6 and 7. The decrease in color intensity is attributed to the oxidation of the dye to its degradable products, results in the formation of formic acid as the main oxidation product.

Investigation of assay parameters

1. Effect of acid type and concentration

The oxidation of Terbinafine HCl, Telmisartan, and Ramipril by ceric sulphate were performed in acid medium. Acid medium is needed to prevent precipitation of the hydrated cerium (IV) oxide ($\text{CeO}_2 \cdot x\text{H}_2\text{O}$). In order to determine the most appropriate acid, different acids (sulphuric, hydrochloric, nitric, perchloric and acetic) were tested, sulphuric acid gave the highest readings.

Using different sulphuric acid concentrations ranging from 0.5-5M solutions, the optimum concentrations that gave maximum decrease in absorbance were 0.5M for Terbinafine HCl, 5M for Telmisartan and 1M for Ramipril.

2. Effect of ceric (IV) sulphate concentration and volume

Maximum increase in absorbance was obtained using 1ml 0.2% w/v ceric (IV) sulphate for Terbinafine HCl, 1ml 0.3% w/v ceric (IV) sulphate for Telmisartan and 1.2 ml 0.4% w/v was the best for Ramipril.

3. Effect of temperature and heating time

Heating for 20min in a boiling water bath was optimum for Terbinafine HCl and 35 min in a boiling water bath was optimum for Telmisartan and Ramipril.

4. Effect of chromotrope 2R (C2R) concentration and volume

Maximum increase in the absorbance was achieved by using 1.4, 1, 1.6ml of 0.15% w/v of C2R for Terbinafine HCl, Telmisartan and Ramipril, respectively.

5. Effect of diluting solvent

The effect of diluting solvent on the absorption intensity of the oxidation reaction of Terbinafine HCl, Telmisartan and Ramipril with the different reagents was studied using various solvents for dilution (water, methanol, ethanol, acetonitrile, acetone and isopropanol). It was found that water was the optimum diluting solvent for the studied drug as it gave maximum readings. Therefore water was selected for further work with all reagents.

VII. Validation of the proposed methods

Linearity

The methods were tested for linearity, accuracy and precision. By using the above procedures, linear regression equations were obtained. The regression plots showed a linear dependence of the absorbance over Beer's law range given in Table 1. The table also shows the results of the statistical analysis of the experimental data,

such as the slopes, the intercepts, the correlation coefficients obtained by the linear least-squares treatment of the results and Molar absorptivity. Results of recovery studies with pure drugs by proposed methods (Table 2) show small values of standard deviation and variance that indicates low scattering of the points around the calibration line and high precision.

Limit of quantitation and limit of detection

The limits of quantitation (LOQ) were determined by establishing the lowest concentration that can be measured according to ICH recommendation [49] below which the calibration graph is non linear. The results are shown in Table 1. The limits of detection (LOD) were determined by evaluating the lowest concentration of the analyte that can be readily detected. The results are also summarized in Table 1.

Precision

The precisions of the assays (intra-day and inter-day) were determined for the studied drugs concentrations cited in Table 3. The assays, gave satisfactory results (Table 3). This level of precision of the proposed methods was adequate for the quality control analysis of TER, TEL and RP.

Accuracy

The results obtained were in good agreement with those obtained using the reference

methods (50, 51 and 52). Statistical analysis of the result obtained using student t-test and the variance ratio F-test revealed no significance differences between the proposed and references methods regarding the accuracy and precision, respectively (Table 4).

Analytical applications

The results obtained by applying the proposed methods for the determination of drugs in their pharmaceutical formulations (Lamisil, Micardis and Tritace protect tablets) (Table 5, 6) suggest satisfactory recovery. Further, standard addition technique followed to check the validity of the method has given good recoveries of the drugs. Hence, these methods can be recommended for adoption in routine analysis of TEL, TER and RP.

VIII. CONCLUSION

The proposed methods are sensitive, enabling accurate and precise determination of TH, TEL and RP over satisfactory concentration ranges without the need for special or laborious sample-pretreatment steps. The methods which are advantageously time and cost-efficient are successfully applied for the quantification of the drugs in commercial samples, with results being in good statistical agreement with reference method. Therefore, the proposed methods are considered useful for routine quality monitoring of pharmaceuticals.

Table 1: Spectral data for determination of Terbinafine HCl, Telmisartan and Ramipril using Cerium (IV) sulphate and cerium (IV) sulphate-chromotrope 2R. **A=a + b c

Parameters	Cerium(IV) sulphate method			Cerium(IV) sulphate + C2R method		
	Terbinafine HCl	Telmisartan	Ramipril	Terbinafine HCl	Telmisartan	Ramipril
Linearity range ($\mu\text{g ml}^{-1}$)	1-7	1-5	10-70	1-7	1-5	10-50
Wavelength (nm)	319	319	319	507	507	507
Limit of detection (LOD) ($\mu\text{g ml}^{-1}$)	1.06	1.33	1.02	0.93	1.43	0.7
Limit of quantification (LOQ) ($\mu\text{g ml}^{-1}$)	3.53	4.44	3.40	3.11	4.78	2.12
Regression equation**:						
Slope (b)	0.1486	0.1736	0.0153	0.1634	0.226	0.0221
Intercept (a)	0.0059	0.0328	-0.0034	-0.0451	0.028	0.029
Correlation coefficient (r)	0.9999	0.9995	0.9998	0.9997	0.9997	0.9999
SE	0.33	0.44	0.28	0.31	0.52	0.2
Reproducibility (R. S. D %)	1.27	0.64	0.94	0.67	0.51	0.99
Repeatability (R. S. D %)						
Molar absorptivity ($\text{L mol}^{-1}\text{cm}^{-1}$)	1.38	1.1	1.2	1.6	0.69	1.2
	1.5×10^3	1.8×10^3	1.5×10^3	1.6×10^3	2.4×10^3	2.3×10^3

Table 2: Determination of Terbinafine HCl, Telmisartan and Ramipril using cerium(IV) sulphate and cerium(IV) sulphate-chromotrope 2R

	Cerium(IV) sulphate method						Cerium(IV) sulphate + C2R method					
	Terbinafine HCl		Telmisartan		Ramipril		Terbinafine HCl		Telmisartan		Ramipril	
	Taken $\mu\text{g ml}^{-1}$	Recovery %	Taken $\mu\text{g ml}^{-1}$	Recovery %	Taken $\mu\text{g ml}^{-1}$	Recovery %	Taken $\mu\text{g ml}^{-1}$	Recovery %	Taken $\mu\text{g ml}^{-1}$	Recovery %	Taken $\mu\text{g ml}^{-1}$	Recovery %
1	98.32		1	98.62	10	100.26	1	99.39	1	98.23	10	100.00
2	98.96		1.5	99.92	30	98.78	2	99.36	2	100.00	20	99.77
3	100.96		3.5	99.93	40	100.23	3	101.67	3	100.59	30	99.70
4	100.29		4	101.84	60	100.59	4	100.4	4	100.88	40	100.79
5	100.28		4.5	99.49	70	99.29	5	100.8	5	99.29	50	99.64
6	100.39		5	99.33			6	100				
7	99.41						7	99.3				
Mean \pm S.D.	99.8 \pm 0.93		99.86 \pm 1.09		99.83 \pm 0.76		100.13 \pm 0.823		99.8 \pm 1.07		99.98 \pm 0.47	
N	7		6		5		7		5		5	
S.D.	0.93		1.09		0.76		0.823		1.07		0.47	
R.S.D.	0.94		1.09		0.76		0.82		1.07		0.47	
V	0.87		1.2		0.58		0.79		1.14		0.18	
S.E.	0.35		0.44		0.34		0.31		0.48		0.21	

* Average of three experiments

Table 3: Precision of the proposed methods for analysis of Terbinafine HCl, Telmisartan and Ramipril

	Drug	Terbinafine HCl	Telmisartan	Ramipril	Terbinafine HCl	Telmisartan	Ramipril
	Method	Cerium (IV) sulphate method	Cerium (IV) sulphate + C2R method				
	Wavelength (nm)	319 nm	319 nm	319 nm	507 nm	507 nm	507 nm
	Weight taken ($\mu\text{g/ml}$)	2	1.5	40	3	1	10
	Validation Parameters						
	%Recovery						
	Experiment						
Intra-day	1	99	100	99	101	101	102
	2	102	101	97	102	100	100
	3	101	100	99	100	100	101
	4	99	100	100	100	100	102
	5	100	101	100	100	100	102
	Mean	100	101	99	101	101	101
	S.D.	1.28	0.64	1.18	0.67	0.51	1
	R.S.D.	1.27	0.64	1.19	0.67	0.51	0.99
Inter-day	%Recovery						
	Experiment						
	1	99	98	99	99	98	99
	2	100	101	97	98	97	98
	3	99	100	99	101	97	100
4	102	100	99	99	98	99	
	Mean	100	99	98	99	98	99
	S.D.	1.38	1.1	0.92	1.59	0.68	1.1
	R.S.D.	1.38	1.1	.94	1.6	0.69	1.2

Table 4: Statistical data for determination of Terbinafine HCl, Telmisartan and Ramipril using ceric (IV) sulphate and ceric (IV) sulphate-chromotrope 2R

Drug	Parameters	ceric(IV) sulphate method	ceric(IV) sulphate+ C2R method	Reference method
Terbinafine HCl	Mean \pm S.D	99.80 \pm 0.93	100.13 \pm 0.82	100.60 \pm 0.96 ^[50]
	N	7	7	5
	Variance	0.87	0.79	0.92
	Student-t-test	1.45 (2.228)*	0.87 (2.228)*	
	F-test	1.06 (4.53)*	1.16 (4.53)*	
Telmisartan	Mean \pm S.D	99.86 \pm 1.1	99.80 \pm 1.07	99.99 \pm 1.02 ^[51]
	N	6	5	4
	Variance	1.2	1.14	1.03
	Student-t-test	0.19 (2.306)*	0.27 (2.365)*	
	F-test	1.17 (6.41)*	1.11 (6.69)*	
Ramipril	Mean \pm S.D	99.83 \pm 0.76	99.98 \pm 0.47	100.17 \pm 0.86 ^[52]
	N	5	5	5
	Variance	0.58	0.18	0.74
	Student-t-test	0.66 (2.306)*	0.44 (2.306)*	
	F-test	1.28 (6.39)*	4 (6.39)*	

*Tabulated values of t and F at p = 0.05

Table 5: Application of standard addition technique for determination of Terbinafine HCl, Telmisartan and Ramipril in their pharmaceutical formulations using cerium(IV) sulphate

	Terbinafine HCl (Lamisil tablets)			Telmisartan (Micardis tablets)			Ramipril (Tritace Protect tablets)		
	Taken	Added	Recovery*	Taken	Added	Recovery*	Taken	Added	Recovery*
	$\mu\text{g ml}^{-1}$		%	$\mu\text{g ml}^{-1}$		%	$\mu\text{g ml}^{-1}$		%
	1	--	98.32	1	--	99.19	10	--	100.26
		1	98.99		0.5	99.31		20	99.15
		2	98.28		1	99.77		30	100.96
		4	97.93		1.5	98.77		50	99.79
		5	97.19		2	99.42		60	98.41
					3	98.54			
					3.5	99.11			
Mean \pm S.D.	98.4 \pm 0.66			99 \pm 0.41			99.71 \pm 0.98		
N	4			6			4		
V	0.43			0.17			0.97		
S.D.	0.66			0.41			0.98		
S.E.	0.33			0.17			0.49		

* Mean of three different experiments

Table 6: Application of standard addition technique for determination of Terbinafine HCl, Telmisartan and Ramiprii in their pharmaceutical formulations using cerium(IV) sulphate-chromotrope 2R

	Terbinafine HCl (Lamisil tablets)			Telmisartan (Micardis tablets)			Ramipril (Tritace Protect tablets)		
	Taken	Added	Recovery*	Taken	Added	Recovery*	Taken	Added	Recovery*
	$\mu\text{g ml}^{-1}$		%	$\mu\text{g ml}^{-1}$		%	$\mu\text{g ml}^{-1}$		%
	2	--	99.48	1	--	98.24	10	--	97.74
		2	99.48		1	97.36		10	97.74
		3	98.96		1.5	97.80		20	96.38
		4	98.70		3	100.29		25	97.01
		5	100.99		4	98.35		30	95.93
Mean \pm S.D.	99.52 \pm 0.89			97.96 \pm 0.52			96.96 \pm 0.81		
N	4			4			4		
V	0.79			0.27			0.65		
S.D.	0.89			0.52			0.81		
S.E.	0.44			0.26			0.4		

* Mean of three different experiments

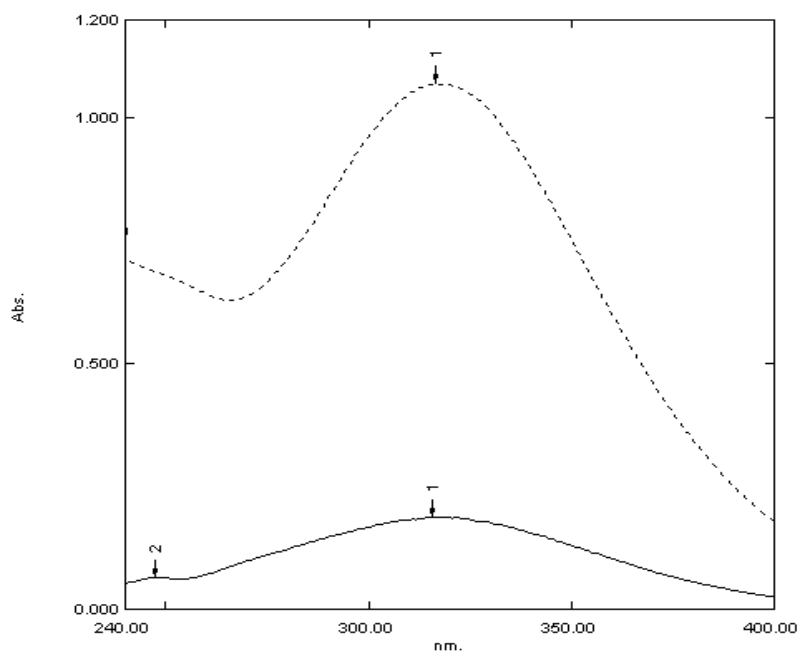


Fig. 2: Absorption spectra of the reaction of Ce (IV) sulphate with $50 \mu\text{g ml}^{-1}$ Terbinafine HCl (—) and Blank (- - - -)

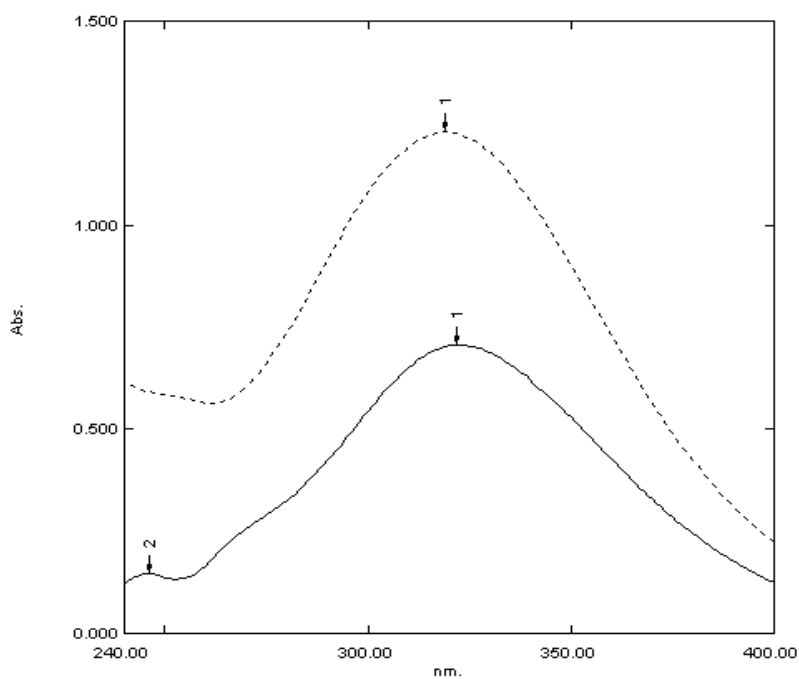


Fig. 3: Absorption spectra of the reaction of Ce (IV) sulphate with $100 \mu\text{g ml}^{-1}$ Telmisartan (—) and Blank (- - - -)

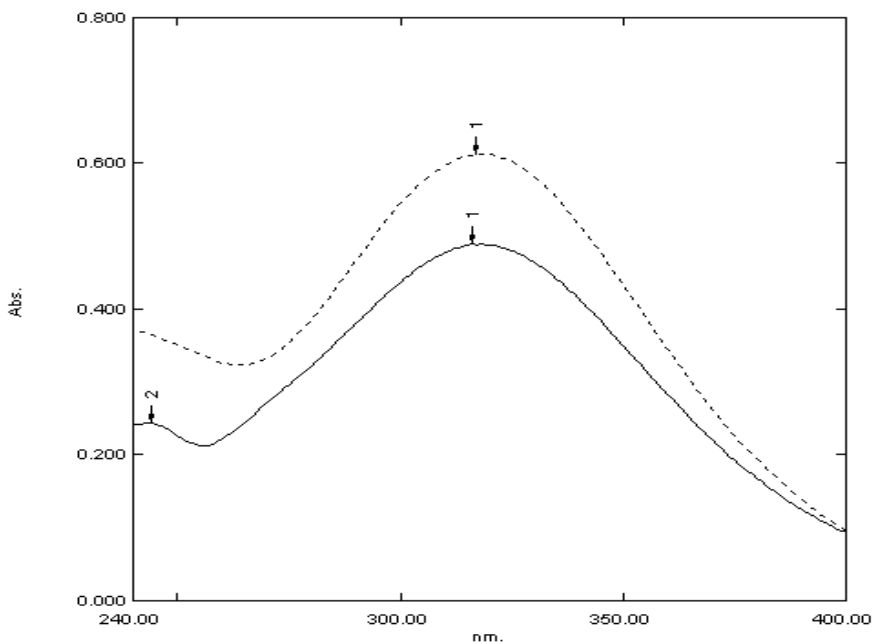


Fig. 4: Absorption spectra of the reaction of Ce(IV) sulphate with 50 µg ml⁻¹ Ramipril (—) and Blank (---)

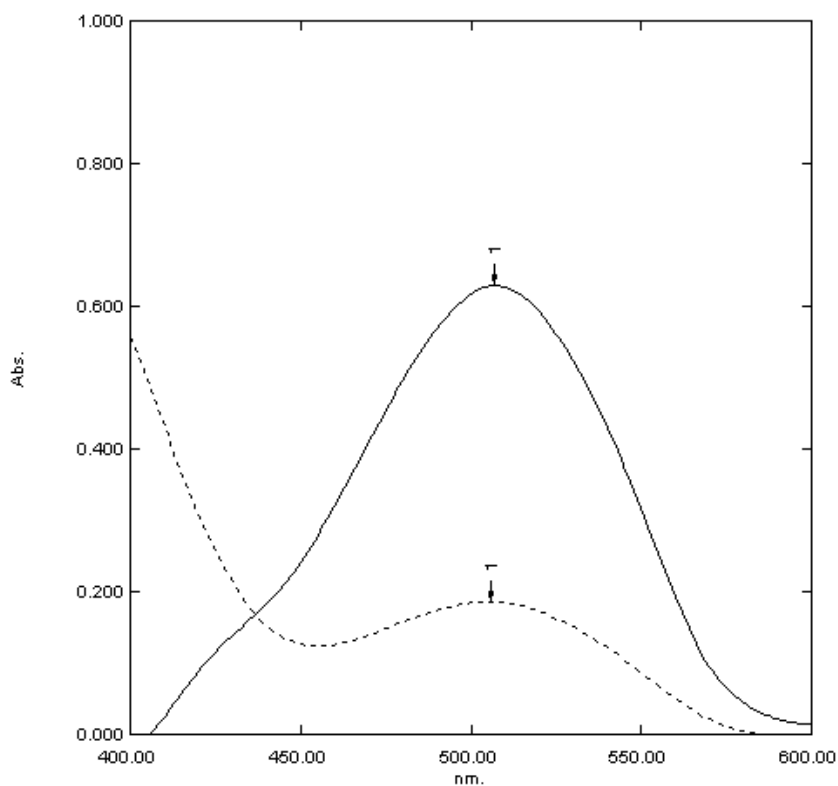


Fig. 5: Absorption Spectra of the reaction products of Cerium (IV) sulphate and C2R with 50 µg ml⁻¹ Terbinafine HCl (—) and Blank (---)

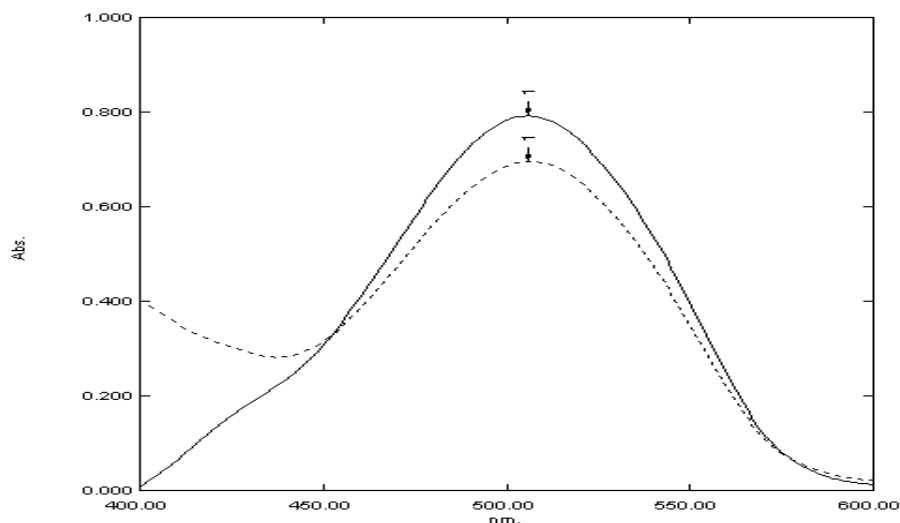


Fig. 6: Absorption Spectra of the reaction products of Cerium (IV) sulphate and C2R with $100 \mu\text{g ml}^{-1}$ Telmisartan (—) and Blank (----)

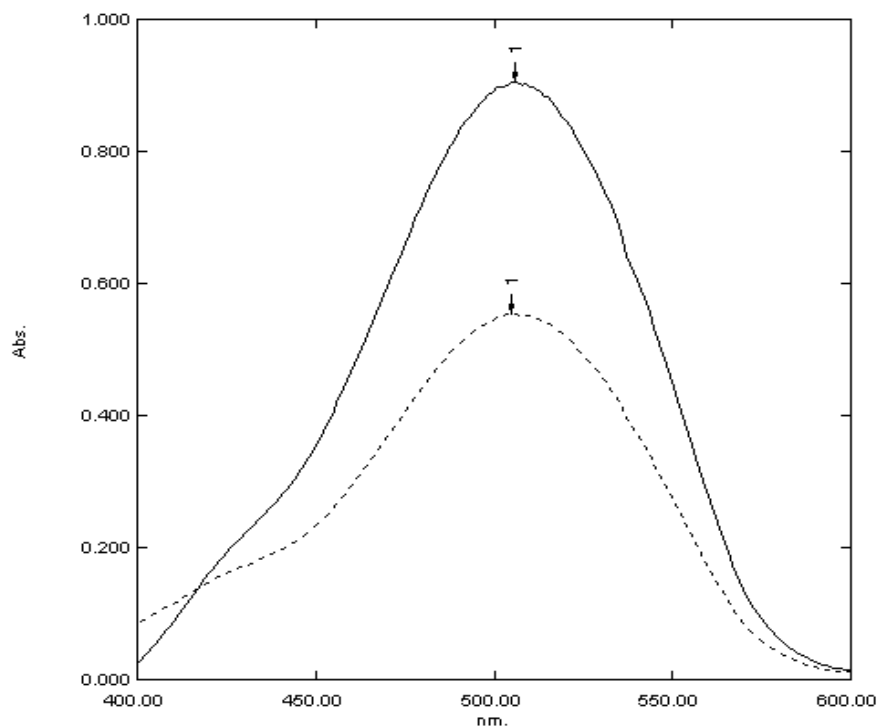


Fig. 7: Absorption Spectra of the reaction products of Cerium (IV) sulphate and C2R with $50 \mu\text{g ml}^{-1}$ Ramipril (—) and Blank (----)

REFERENCES

1. British Pharmacopeia. 2007;22000-2001.
2. European Pharmacopoeia. 2011;2:3024-3025.
3. Block JH and Beale JM. Wilson and Gisvold's. Textbook of Organic and Pharmaceutical chemistry. 11th edition, Published by Lippincott Williams and Wilkins. 2004;239.

4. Patel KK and Kakhanis VV. A validated HPTLC method for determination of Terbinafine hydrochloride in pharmaceutical solid dosage form. *International Journal of Pharmaceutical Sciences and Research*. 2012;3(11):4492-4495.
5. Ahmad S, Jain GK, Faiyazuddin M, Iqbal Z, Talegaonkar S, Sultana Y and Ahmad FJ. Stability-indicating high-performance thin-layer chromatographic method for analysis of terbinafine in pharmaceutical formulations. *Acta Chromatogr*. 2009;21(4):631-639.
6. Suma BV, Kannan K, Madhavan V and Nayar CR. HPTLC Method for determination of Terbinafine in the Bulk drug and Tablet dosage form. *International Journal of Chem Tech Research*. 2011;3(2):742-748.
7. Wang C, Mao Y, Wang D, Yang G, Qu Q and Hu X. Voltammetric determination of Terbinafine in biological fluid at glassy carbon electrode modified by cysteic acid/carbon nanotubes composite film. *J bioelectrochem*. 2008;72(1):107-115.
8. Goswami PD. Validated spectrophotometric method for the estimation of Terbinafine hydrochloride in bulk and in tablet dosage form using inorganic solvent. *Der Pharmacia Lettre*. 2013;5(3):386-390.
9. Abdel-Moety EM, Kelani KO and Abou al-Alamein AM. Spectrophotometric determination of terbinafine in presence of its photodegradation products. *Boll Chim Farm*. 2002;141(4):267-273.
10. Patel KK, Marya BH and Kakhanis VV. Spectrophotometric determination and validation for Terbinafine Hydrochloride in pure and in tablet dosage form. *Der Pharmacia Lettre*. 2012;4(4):1119-1122.
11. Jain PS, Chaudhary AJ, Patel SA, Patel ZN and Patel DT. Development and validation of the UV spectrophotometric method for determination of Terbinafine hydrochloride in bulk and in formulation. *Pharmaceutical methods*. 2011;3(2):198-202.
12. Cardoso SG and Schapoval EES. UV spectrophotometry and nonaqueous determination of Terbinafine hydrochloride in dosage forms. *Journal of AOAC International*. 1999;82(4):830-833.
13. Patel KK and Kakhanis VV. A validated HPTLC method for determination of Terbinafine hydrochloride in pharmaceutical solid dosage form. *International Journal of Pharmaceutical Sciences and Research*. 2012;3(11):4492-4495.
14. Abdel-Moety EM, Kelani KO and Abou Al-Alamein AM. Chromatographic determination of Terbinafine in presence of its photodegradation products. *Saudi pharmaceutical journal*. 2003;11(1-2):37-45.
15. Matysova L, Solich P, Marek P, Havlikova L, Novakova L and Sicha J. Separation and determination of Terbinafine and its four impurities of similar structure using simple RP-HPLC method. *Talanta*. 2006;68(3):713-720.
16. Denouel J, Keller HP, Schaub P, Delaborde C and Humbert H. Determination of Terbinafine and its desmethyl metabolite in human plasma by high-performance liquid chromatography. *Journal of Chromatography*. 1995;663(2):353-359.
17. De Oliveira CH, Barrientos-Astigarraga RE, De Moraes MO, Bezerra FA, De Moraes ME and De Nucci G. Terbinafine quantification in human plasma by high-performance liquid chromatography coupled to electro spray tandem mass spectrometry: application to a bioequivalence study. *Therapeutic drug monitoring*. 2001;23(6):709-716.
18. Kuznets J, KoruEre N and Drobnic-Kosorok M. Determination of Terbinafine HCl in cat hair by two chromatographic methods. *Biomedical chromatography*. 2001;15(8):497-502.
19. Patel KK. A validated RP-HPLC method for determination of Terbinafine Hydrochloride in pharmaceutical solid dosage form. *International Journal of Pharmacy and Technology*. 2012;4(3):4663-4669.
20. Cardoso GS and Schapoval EES. High performance liquid chromatographic assay of Terbinafine hydrochloride in tablets and creams. *Journal of Pharmaceutical and Biomedical Analysis*. 1999;19: 809-812.
21. Banked S, Tapadiya GG, Saboo SS, Bindaiya S, Deepti Jain and Khadbadi SS. Simultaneous Determination of Ramipril, Hydrochlorothiazide and Telmisartan by Spectrophotometry.

- Inter J of Chem Tech Research. 2009;1:183-188.
22. Patil UP, Gandhi SV, Sengar MR and Raj mane VS. Simultaneous Determination of Atorvastatin Calcium and Telmisartan in Tablet Dosage Form by Spectrophotometry. International Journal of Chem Tech Research. 2009;1:970-973.
 23. Popat B Mohitea, Ramdas B Pandharea and Vaidhun H Bhaskar. Eurasian. Simultaneous Estimation of Ramipril and Telmisartan in Tablet Dosage Form by Spectrophotometry. J Anal Chem. 2010;5:89-94.
 24. Asha B Thomas, Sheetal N Jagdale, Shweta B Dighe and Rabindra K Nanda. Simultaneous Spectrophotometric Estimation of Amlodipine Besylate and Telmisartan in Tablet Dosage Form. Int J PharmTech Res. 2010;2:1334-1341.
 25. Vekaria NR, Fursule RA and Surana SJ. Application of UV-spectrophotometry and First Order Derivative Methods for Determination of Telmisartan in bulk and tablets. Orient J Chem. 2008; 24(1):353-356.
 26. Zonghui Qin, Weifen Niu and Ron Tan. Spectrophotometric method for the determination of Telmisartan with Congo red. J Anal Chem. 2009;64:449-454.
 27. Sunil Jawla, Jeyalakshmi K, Krishnamurthy T and Kumar Y. Development and Validation of Simultaneous HPLC method for Estimation of Telmisartan and Ramipril in Pharmaceutical Formulations. Int J PharmTech Res. 2011;2:1625-1633.
 28. Vijayamirharaj R, Ramesh J, Jayalakshmi B and Hanas Bin Hashim. Development and Validation of RP-HPLC Method for the Simultaneous Estimation of Telmisartan and Atorvastatin Calcium in Tablet Dosage Forms. International Journal of Comprehensive Pharmacy(IJCP). 2010;4(03):1-4.
 29. Kottai Muthu A, Sankhla R, Gupta SH, Smith AA and Manavalan R. Development and validation of a reversed Phase HPLC method for simultaneous determination of Amlodipine and Telmisartan in pharmaceutical dosage form. J Chem Res S. 2010;12:43-52.
 30. Gupta A, Charde RM and Charde MS. Determination of Telmisartan and forced degradation behavior by RP-HPLC in tablet dosage Form. Journal of Pharmacy Research. 2011;4(4):1270.
 31. Vinit Chavhan, Rohini Lawande, Jyoti Salunke, Minal Ghante and Supriya Jagtap. UV Spectrophotometric method development and validation for Telmisartan in bulk and tablet dosage form. Asian J Pharm Clin Res. 2013;6(4):19-21.
 32. Gupta VK, Rajeev Jain, Ojitkumar Lukram, Shilpi Agarwal and Ashish Dwivedi. Simultaneous determination of Ramipril, Ramiprilat and Telmisartan in human plasma using liquid chromatography tandem mass spectrometry. Talanta. 2011;83:709-716
 33. Patel VA. Development and Validation of HPTLC Method for the Simultaneous Estimation of Telmisartan and Ramipril in Combined Dosage Form. International Journal of Pharmaceutical and Biological Research. 2010;1(1):18-24.
 34. Lories I Bebawy, Samah S Abbas, Laila A Fattah and Heba H. Refaat. Application of first-derivative, ratio derivative spectrophotometry, TLC-densitometry and spectrofluorimetry for the simultaneous determination of Telmisartan and Hydrochlorothiazide in pharmaceutical dosage forms and plasma. Farmaco. 2005;60:859-867.
 35. Yusuf S, Sleight P and Pogue J. Effects of an angiotensin- converting-enzyme inhibitor, Ramipril, on cardiovascular events in high-risk patients. N Engl J Med. 2000;342:145-153.
 36. Arnold JM, Yusuf S and Young and et al J. Prevention of heart failure in patients in the heart outcomes prevention evaluation (HOPE) study. Circulation. 2003;107:1284-1290.
 37. Dagenais GR, Yusuf S and Bourassa MG. Effects of Ramipril on coronary events in high-risk persons: results of the heart outcomes prevention evaluation study. Circulation. 2001;104:522-526.
 38. Hogan BL, Williams M and Idiculla A. Development and validation of aliquid chromatographic method for the determination of the related substances of Ramipril in Altace capsules. J Pharm Biomed Anal. 2000;23:637-651.
 39. Bonazzi JD, Gotti R and Andrisano V. Analysis of ACE inhibitors in pharmaceutical dosage forms by derivative UV spectroscopy and liquid

- chromatography (HPLC). *J Pharm Biomed Anal.* 1997;16:431-438.
40. Vachareau A and Neirinck L. Liquid chromatography mass spectrometry method for determination of Ramipril and its active metabolite Ramiprilat in human plasma. *J Chromatogr.* 2002;792: 97-306.
41. Kamath BV and Shivram K. Spectrophotometric methods for determination of Diclofenac sodium. *Analytical Letters.* 1993;26(5):903-911.
42. HD, Manju BG and Ramappa PG. Spectrophotometric methods for determination of ritodrine hydrochloride and amoxicillin. *Revanasiddappa. Analytical Science.* 1999;15(7):661-664.
43. Salem H, Saleh and GA. Spectrophotometric methods for determination of phenolic beta-lactam antibiotics. *J Pharm Biomed Anal.* 2002;28(6):1205-1213.
44. Taha EA and Youssef NF. Risedronate sodium, Alendronate sodium and Etidronate disodium. *Chemical Pharmaceutical Bulletin.* 2003;51(12):1444 - 1447.
45. Abdel-Razak O. Spectrofluorimetric and spectrophotometric methods for determination of Ritodrine hydrochloride. *J Pharm Biomed Anal.* 1998;18(3):359-365.
46. El-Walily AFM, Gazy AA, Belal SF and Khamis EF. Spectrofluorimetric and spectrophotometric methods for determination of penicillins, cephalosporins. *Spectroscopy Letters.* 2000;33(6):931-948.
47. Mahgoub H. Spectrofluorimetric and spectrophotometric methods for determination of Aztreonam. *J Pharm Biomed Anal.* 2003;31(4):767-774, 2003.
48. Ayad MM, Abdellatef HE, El-Henawee MM and El-Sayed HM. Spectrofluorimetric and spectrophotometric methods for determination of Acyclovir and Acebutolol hydrochloride. *Spectrochimica Acta A.* 2007;66(1):106-110.
49. ICH Harmonized Tripartite Guideline: Validation of Analytical Procedures. Text and Methodology, Q2 (R1), Current Step 4 Version, Parent Guidelines on Methodology. 2005.
50. Pritam S Jain, Amar J Chaudhari and Dhvani T Patel. Development and validation of the UV-spectrophotometric method for determination of Terbinafine hydrochloride in bulk and in formulation. *Pharm Methods.* 2011;2(3):198-202.
51. Ajit Pandey, Sawarkar H, Mukesh Singh, Kashyap P and Priyanka Ghosh. UV-Spectrophotometric Method for estimation of Telmisartan in Bulk and Tablet Dosage Form. *International Journal of Chem Tech Research.* 2011;3(2):657-660.
52. OE Afieroho, O Okorie and TJN Okonkwo. A Spectrophotometric Method for the Determination of Ramiprilin Solid Dosage Forms. *Tropical Journal of Pharmaceutical Research.* 2012;11(2):275-279.