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Research Article

NEWER APPLICATIONS OF BENZOTHIAZEPINES

AS POTENTIAL CYTOTOXIC AGENTS

Vudumula Kotireddy* and K. Venkata Ramana

Department of pharmaceutical chemistry, A.S.N Pharmacy college, Burripalem Road, Tenali, Andhra Pradesh, India.

ABSTRACT

Benzothiazepines are synthesized by conventional synthesis method. The compounds have been screened for Cytotoxic activity. Tested compounds were prepared by the reactions between 1,3-diarylprop-2-enones with orthoamino thio phenol. All the products were tested for purity by TLC and charecterised by elemental analysis and different spectroscopic methods.

Keywords: 4-Chloroacetophenone, Benzothiazepine, 2-Aminothiophenol, piperidine.

INTRODUCTION

The benzothiazepines are important nitrogensulfur-containing seven-membered and heterocyclic compounds in drug research since possess thev diverse bioactivities. Benzothiazepines are the most well-known representatives of benzologs of 1, 4-thiazepine and one of the three possible benzo-condensed derivatives, viz. 1,4-and 1,5benzothiazepines.The benzothiazepine derivatives are of particular interest for lead discovery because they have been found active against different families of targets. The first molecule of benzothiazepine used clinically was diltiazem, followed by clentiazem, for their cardiovascular action. Therefore. the benzothiazepines are useful compounds in the drug research which has stimulated the invention of a wide range of synthetic methods

for their preparation and chemical transformations.

MATERIALS AND METHODS

Procedure for Synthesis of 1, 5-Benzothiazepines

Chalcones of 4-Chloroacetophenone (1 mill mole) and O-Amino thiophenol (1 mill mole) was dissolved in 10 ml of boiling methanol the heat was removed and piperidine (2 drops) was added. After the mixture had cooled to room temperature the additional 10 ml of methanol was added and heated until the slurry was dissolved. Then add 1 ml of Glacial acetic acid and allow the mixture at 250C for overnight. The yellow color crystals benzothiazepine was separated out. This was recrystallized with methanol and filtered. The scheme and physical characterization data will be given below.

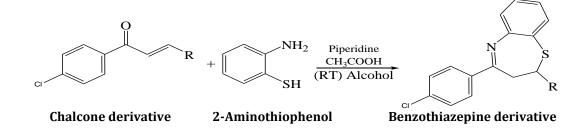


Table 1: Physical characterization data of benzothiazepines (BP1-BP12)						
Compound	R	Molecular Formula	Relative Molecular Mass (RMM)	Melting Point (°C)	Yield %	
BP1		C ₂₂ H ₁₈ ClNS	363	140-143	89	
BP ₂		C ₂₁ H ₁₅ ClFNS	367	153-154	89	
BP ₃		$C_{21}H_{15}Cl_2NS$	383	143-145	93	
BP4		C21H15Cl2NS	383	120-123	71	
BP5	F 	C ₂₁ H ₁₄ ClF ₂ NS	385	138-141	75	
BP ₆		C21H14Cl3NS	416	117-120	86	
BP ₇		C ₂₁ H ₁₄ Cl ₂ N ₂ O ₂ S	428	164-167	77	
BP ₈		C21H15ClN2O2S	394	142-145	82	
BP9		C21H15ClN2O2S	394	130-131	89	
BP ₁₀	ОН	C21H16CINOS	365	226-229	84	
BP ₁₁		C22H17CIN2O2S	408	176-179	94	
BP ₁₂	OCH ₃ —OCH ₃ —OCH ₃	C24H22CINO3S	439	148-151	85	

Table 1: Phy	vsical characterization data of b	enzothiazepines (BP ₁ -BP ₁₂)

Compound	Position of absorption band (cm ⁻¹)		
BP1	1585 (C=N), 1505 (C=C), 1395 (C-N), 823 (C-Cl) and 654 (C-S).		
BP ₂	1625 (C=N), 1509 (C=C),1399 (C-N), 689 (C-S) and 831 (C-Cl)		
BP ₃	1595 (C=N), 1502 (C=C), 1384 (C-N), 778 (C-Cl), 821 (C-Cl) and 667 (C-S)		
BP ₄	1596 (C=N), 1510 (C=C), 1365 (C-N), 688 (C-S), 823 (C-Cl) and 805 (C-Cl)		
BP ₅	1612 (C=N), 1501 (C=C),1382 (C-N), 689 (C-S), 813 (C-Cl) and 944 (C-F)		
BP ₆	1593 (C=N), 1502 (C=C), 1382 (C-N), 687 (C-S) and 925 (C-Cl)		
BP ₇	1588 (C=N), 1520 (N=O, asymmetric), 1505 (C=C), 1382 (C-N), 1340 (N=O, symmetric), 656 (C-S) and 833 (C-Cl)		
BP ₈	1580 (C=N), 1522 (N=O, asymmetric), 1501 (C=C), 1385 (C-N), 1345 (N=O, symmetric), 824 (C-Cl) and 689 (C-S)		
BP9	1586 (C=N), 1515 (N=O, asymmetric), 1506 (C=C), 1380 (C-N), 1338 (N=O, symmetric), 825 (C-Cl) and 713 (C-S)		
BP ₁₀	1653 (C=N), 1528 (C-N), 1502 (C=C), 825 (C-Cl) and 694 (C-S)		
BP ₁₁	1642 (C=N), 1548 (N=O, asymmetric), 1510 (C=C), 1380 (C-N), 1338 (N=O, symmetric), 827 (C-Cl) and 668 (C-S)		
BP ₁₂	1648 (C=N), 1505 (C=C), 1365 (C-N), 1225 (-O-CH ₃), 823 (C-Cl) and 678 (C-S)		

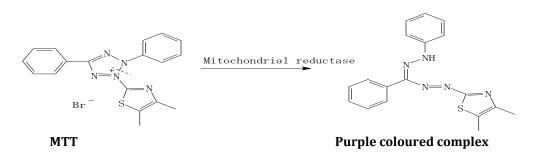
Table 2: IR spectral data (KBr disc) of benzothiazepines (BP₁-BP₁₂)

Table 3: ¹ H NMR spectral data of benzothiazepines (BP₁ - BP₁₀)

Compound	Chemical shift (δ) in ppm
BP1	4.94 (dd, <i>J2</i> ,3a = 5.1 Hz, <i>J2</i> ,3b = 12 Hz, 1H, C2-H), 3.25 (dd, <i>J3a</i> ,3b = 14.4 Hz, <i>J3a</i> ,2 = 9.9 Hz, 1H, C3-H-3a), 3.04 (t, <i>J3b</i> ,3a =
DP1	<i>J3b,2</i> = 12.9 Hz, 1H, C3-H-3b), 2.40 (3H, s, Ar-CH3), 7.22 (1H, s, Ar-H), 7.61 (3H, m, Ar-H), 7.20-8.10 (7H, Ar-H).
BP2	5.27 (dd, <i>J2</i> ,3 <i>a</i> = 5.1 Hz, <i>J2</i> ,3 <i>b</i> = 12 Hz, 1H, C2-H), 3.50 (dd, <i>J3a</i> ,3 <i>b</i> = 14.4 Hz, <i>J3a</i> ,2 = 9.6 Hz, 1H, C3-H-3a), 2.97 (t, <i>J3b</i> ,3 <i>a</i> =
DP2	<i>J3b,</i> 2 = 12.9 Hz, 1H, C3-H-3b), 7.05 (1H, s, Ar-H), 7.19 (3H, m, Ar-H), 7.20-8.09 (7H, Ar-H).
BP3	5.0 (dd, <i>J2</i> ,3 <i>a</i> = 5.1 Hz, <i>J2</i> ,3 <i>b</i> = 12 Hz, 1H, C2-H), 3.53 (dd, <i>J3a</i> ,3 <i>b</i> = 14.4 Hz, <i>J3a</i> ,2 = 9.9 Hz, 1H, C3-H-3a), 3.39 (t, <i>J3b</i> ,3 <i>a</i> =
DP3	<i>J3b,2</i> = 12.9 Hz, 1H, C3-H-3b), 7.25 (1H, s, Ar-H), 7.65 (3H, m, Ar-H), 7.22-8.08 (7H, Ar-H).
BP4	4.89 (dd, <i>J2,3a</i> = 5.1 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 3.43 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.6 Hz, 1H, C3-H-3a), 3.36 (t, <i>J3b,3a</i> =
DF4	<i>J3b,2</i> = 12.9 Hz, 1H, C3-H-3b), 7.12 (1H, s, Ar-H), 7.72 (3H, m, Ar-H), 6.95-7.60 (7H, Ar-H).
BP5	5.31 (dd, <i>J2,3a</i> = 5.1 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 3.36 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.9 Hz, 1H, C3-H-3a), 2.87 (t, <i>J3b,3a</i> =
DIJ	<i>J3b,</i> 2 = 12.9 Hz, 1H, C3-H-3b), 7.08 (1H, s, Ar-H), 7.30 (3H, m, Ar-H), 6.98-8.12 (6H, Ar-H).
BP6	5.10 (dd, <i>J2,3a</i> = 5.1 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 3.27 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.6 Hz, 1H, C3-H-3a), 2.66 (t, <i>J3b,3a</i> =
Dro	<i>J3b,</i> 2 = 12.9 Hz, 1H, C3-H-3b), 7.15 (1H, s, Ar-H), 7.20 (3H, m, Ar-H), 7.05-7.95 (6H, Ar-H).
BP7	4.32 (dd, <i>J2,3a</i> = 5.1 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 3.74 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.9 Hz, 1H, C3-H-3a), 3.51 (t, <i>J3b,3a</i> =
DI /	<i>J3b,2</i> = 12.9 Hz, 1H, C3-H-3b), 7.09 (1H, s, Ar-H), 7.12 (3H, m, Ar-H), 6.98-8.10 (6H, Ar-H).
BP8	5.42 (dd, <i>J2,3a</i> = 5.1 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 3.38 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.6 Hz, 1H, C3-H-3a), 2.86 (t, <i>J3b,3a</i> =
DIO	<i>J3b,</i> 2 = 12.9 Hz, 1H, C3-H-3b), 7.30 (1H, s, Ar-H), 7.80 (3H, m, Ar-H), 7.48-8.60 (7H, Ar-H).
BP9	5.42 (dd, <i>J2,3a</i> = 5.1 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 3.47 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.7 Hz, 1H, C3-H-3a), 3.10 (t, <i>J3b,3a</i> =
DI 9	<i>J3b,2</i> = 12.9 Hz, 1H, C3-H-3b), 7.18 (1H, s, Ar-H), 7.25 (3H, m, Ar-H), 7.25-8.20 (7H, Ar-H).
BP10	3.85 (dd, <i>J2,3a</i> = 5.1 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 3.34 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.0 Hz, 1H, C3-H-3a), 2.41 (t, <i>J3b,3a</i> =
DI 10	<i>J3b,2</i> = 12.9 Hz, 1H, C3-H-3b), 7.25 (1H, s, Ar-H), 7.30 (3H, m, Ar-H), 7.15-7.80 (7H, Ar-H), 6.85 (1H, s, Ar-OH).
BP11	4.16 (dd, <i>J2,3a</i> = 5.1 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 3.23 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.9 Hz, 1H, C3-H-3a), 2.53 (t, <i>J3b,3a</i> =
	<i>J3b,2</i> = 12.9 Hz, 1H, C3-H-3b), 2.50 (3H, s, Ar-CH3), 7.30 (1H, s,Ar-H), 6.70 (3H, m, Ar-H), 7.45-8.78 (6, Ar-H)
BP12	3.06 (dd, <i>J2,3a</i> = 5.3 Hz, <i>J2,3b</i> = 12 Hz, 1H, C2-H), 2.83 (dd, <i>J3a,3b</i> = 14.4 Hz, <i>J3a,2</i> = 9.9 Hz, 1H, C3-H-3a), 2.0 (t, <i>J3b,3a</i> =
	<i>J3b,2</i> = 12.9 Hz, 1H, C3-H-3b), 7.22 (1H, s,Ar-H), 6.60 (3H, m, Ar-H), 7.30-7.50 (5H, Ar-H), 3.70 (3H, s, Ar-OCH3), 3.88
	(6H, s, 2XAr-OCH3)

CYTOTOXICITY STUDIES

The *in vitro* cytotoxicity of the test compounds (**B1 to B12**) was evaluated by the MTT assay. This is a colorimetric assay that measures the reduction of yellow 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide (MTT) by mitochondrial succinate dehydrogenase. The MTT enters the cells and passess into the mitrochondria where it is reduced to an insoluble, coloured (dark purple) formazan product. The cells are then solubilized with DMSO and the released, solubilized formazan reagent is measured spectrophotometrically at 570 nm. Since reduction of MTT can only occur in metabolically active cells, the level of activity is a measure of the viability of the cells. When the amount of dark purple formazan produced by the cells is treated with a agent compared with the amount of formazan produced by untreated control cells, the effectiveness of the agent in causing death of cells can be deduced through the production of a dose-response curve.



REDUCTION OF MTT MATERIALS

HT-29 (colon cancer), MCF-7 (breast cancer) and DU- 145 (prostate cancer) cell lines were obtained from National Centre for Cell Science (NCCS), Pune, India. DMEM (Dulbeccos Modified Eagels Medium), MEM (Minimum Essential Media Eagle), MTT [3-(4,5- dimethylthiazol-2yl)-2,5-diphenyltetrazolium bromide], Trypsin, EDTA were purchased from Sigma chemicals (St.Louis,MO). Fetal bovine serum (FBS) was purchased from Arrow Labs, 96 well flat bottom tissue culture plates were purchased from Tarson.

METHOD

a) Maintainence of cell lines

HT-29 and DU-145 cell lines were grown as adherent in DMEM media, whereas MCF-7 was grown in MEM media supplemented with 10% fetal bovine serum. The cultured was maintained in a humidified atmosphere with 5% CO2.

b) Preparation of samples for cytotoxicity

Stock solutions of test compounds (B1 to B25) were prepared (10 mg/mL) in DMSO and from them various dilutions were made with sterile water to get the final drug concentrations of 10, 50, 100 and 200 mg/mL.

c) Cytotoxicity evaluation

The cells were seeded in 96 well plates at a density of 1×104 (counted by Tryphan blue exclusion dye method) per well and were incubated for 24 h to recover. After incubation

the medium was replaced with fresh media containing different dilutions of the test compounds. Then the plated were incubated for additional 48 h at 370C in DMEM/MEM with 10% FBS medium. Following incubation, the medium was removed and replaced with 90 µl of fresh DMEM without FBS. To the above wells, 10 µl of MTT reagent (5 mg/mL of stock solution in DMEM without FBS) was added and incubated at 370C for 3-4 h, there after the above media was replaced by adding 200 µl of DMSO to each well (to dissolve the blue formazan crystals) and incubated at 370C for 10 min. The absorbance at 570 nm was measured on a spectrophotometer.

Methotrexate was used as reference drug for comparison. Assay was performed in triplicate for three independent determinations. The cytotoxicity was expressed as IC50 (μ g/mL) which is the concentration of the compound that inhibited proliferation rate of the tumor cells by 50% as compared to the control untreated cells. IC50 values were determined from the plot: % inhibition versus concentration.

% inhibition at the given concentration =

1- (Absorbance average)	
	x 100
(Control absorbance average)	

IC50=Inv.log(50-c) / m; c and m derived from y=mx+c of plot of % inhibition Vs log C. The results of the compounds are shown in table 4.

$(\text{Br}_1 \text{ to Br}_{11}):(105_0 \text{ values in } \mu\text{g/mL})$				
		Cell line		
Compound	R	HT-29	MCF-7	DU-145
BP_1	4"-methyl phenyl	55 ± 2	62 ± 2	52 ± 1
BP ₂	4"-fluorophenyl	42 ± 2	48 ± 1	62 ± 2
BP ₃	4"-chlorophenyl	92 ± 2	78 ± 2	65 ± 2
BP ₄	2"-chlorophenyl	105 ± 2	168 ± 1	122 ± 2
BP ₅	2",4"-difluorophenyl	28 ± 1	42 ± 2	33 ± 2
BP_6	2",4"-dichlorophenyl	42 ± 2	67 ± 1	56 ± 2
BP7	2"-chloro-5"-nitrophenyl	115 ± 2	NA	NA
BP ₈	3"-nitrophenyl	180 ± 2	NA	NA
BP9	4"-nitrophenyl	155 ± 1	NA	105 ± 2
BP10	3"-hydroxyphenyl	148 ± 2	129 ± 2	155 ± 1
BP11	3"-nitro-4"methylphenyl	64 ± 2	58 ± 1	46 ± 2
BP12	3",4",5"-trimethoxyphenyl	132 ± 2	NA	93 ± 2
Methotrexate		11 ± 1	9±1	6 ± 1

Table 4: Cytotoxicity of the new benzothiazepines (BP₁ to BP₁₁):(IC₅₀ values in μg/mL)

CONCLUSION

All the synthesized benzothiazepines have been evaluated for their cytotoxicity against HT-29 (colon cancer), MCF-7 (breast cancer) and DU-145 (prostate cancer) cell lines. Methotrexate was used as the reference standard. The results clearly revealed that most of the 1,5benzothiazepines possessed cytotoxic activity as evidenced by the IC50 values and is much higher than that of the chalcones indicating the positive contribution of benzothiazepine nucleus in enhancing the cytotoxic activity. Infact, a number of anticancer drugs being used currently possessed benzothiazepine nucleus as part of their structures.

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